



# Interactive Science Notebooks

# What are Interactive Science Notebooks?

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- ❑ A student thinking tool
- ❑ And organizer for inquiry questions and what I learned...
- ❑ A way to access and process the learning utilizing various modalities (writing, drawing, and discussion)
- ❑ A place for writing rough drafts based on hands-on learning
- ❑ A formative assessment tool for teachers

# Why are we using Interactive Science Notebooks?

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- Record data
- Study for tests
- Record progress
- Communication

# Science Notebook Supplies

Composition  
Notebook



Glue



pens &  
pencils



scissors



colored pencils



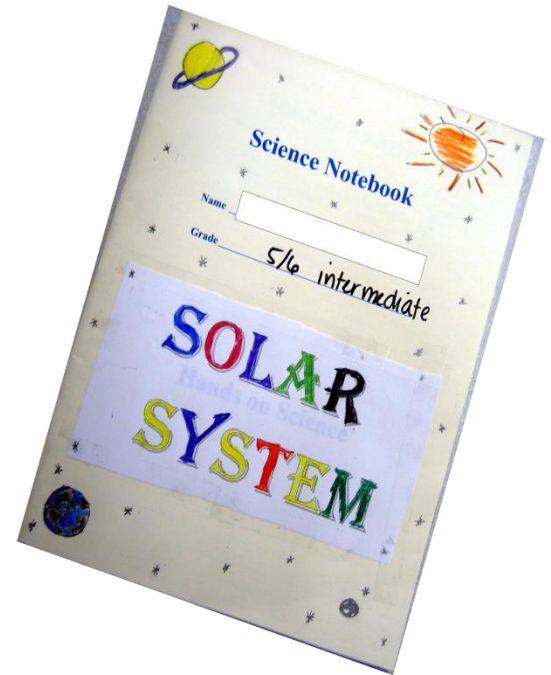
**NO MARKERS!**

# LET'S GET STARTED...

## Cover or Title Page

Give your science notebook a title.

This should give the reader an idea of what this notebook will be about.





## SCIENCE NOTEBOOK EXPECTATIONS AND RUBRIC









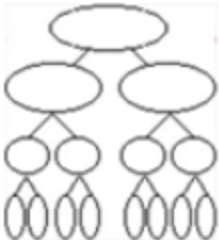



Your Science Notebook is evidence of your hard work and learning, so treat it with utmost care. You are expected to have it with you at all times. Use the posted example of the notebook as a guide for the correct format.

1. Begin each new lesson by adding the title to the top of your next blank page and recording the date in the upper RIGHT corner. Your title can be at the top of either page. Page numbers are always located in the bottom RIGHT corner.
2. If time allows you may add the lesson title and page number to your table of contents, otherwise, you'll need to complete that at the end of the lesson.
3. Each page must have the title, date, and page number. Lessons must be in order and listed in the appropriate table of contents, based on the current six-week period.
4. All papers associated with each lesson must be taped or glued neatly on the page of the activity. No papers should be loose or sticking out from the Notebook.
5. Everything you're asked to write in the course of a lesson should be written in your notebook neatly.
6. Skip lines between the IN activity and the OUT activity.
7. The notebook should be relatively clear of doodling. Use the white space instead for illustrations related to and explaining the science information.
8. The IN Activities must be answered in such a way that you know what the question was asking. For example:
  - a. IN Activity: "Which of the following items are examples of physical characteristics?"
  - b. Appropriate answer: "Color, size, and shape are examples of physical characteristics." This answer will allow you to study from it and understand the content of the lesson later.
  - c. Not an appropriate answer: "Color, size, and shape." When you read this later, you won't have any idea what the lesson was about.
9. Table of Contents needs to be updated regularly with date, lesson title, and page #.
10. AFTER EACH ACTIVITY you should go through your science notebook to:
  - Add color to any diagrams and/or illustrations from the lesson.
  - Ensure that your OUT activity is fully complete and reflective of your learning.

Inside Cover

Student Name  
Ms. Wolfe  
2015-2016

## Science Graphic Organizers

<p><b>T-Chart</b></p> 	<p><b>Box &amp; T-Chart</b></p> 	<p><b>Fish Bones</b></p> 	<p><b>Net</b></p> 
<p><b>Venn Diagram</b></p> 	<p><b>Triple Venn Diagram</b></p> 	<p><b>Wheel</b></p> 	<p><b>Cycle</b></p> 
<p><b>Tree</b></p> 	<p><b>Cause &amp; Effect</b></p> 	<p><b>Chain or Sequence</b></p> 	<p><b>Layer</b></p> 

# Graphic Organizers

- There are twelve graphic organizers that are acceptable to use in this classroom.
- I may ask you to use a specific GO, or sometimes you will be able to choose one.



**Title**

**Date**

*Student Side*

*Teacher Side*

**(1) IN activity**

Essential Question

Bell Ringer

Quick-Write

**(3) OUT activity**

Exit Ticket

Quick-Write

Graphic Organizer

**(2) THROUGH activity**

Lesson Notes

Diagrams/Illustrations

Lab Sheets

Data and Graphs

**Pg. #**

## Table of Contents

Date

Title

Page

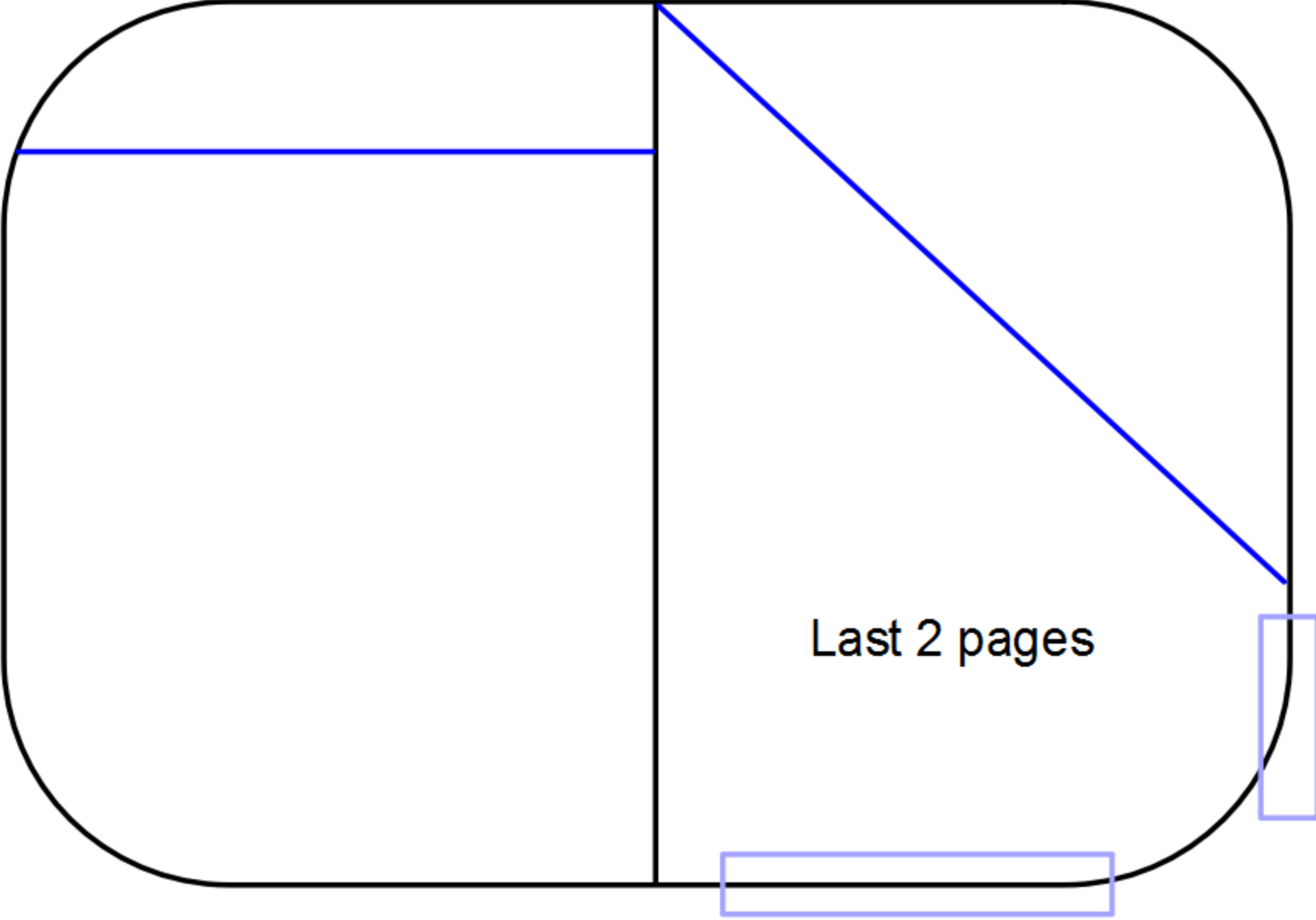
## Table of Contents

Date

Title

Page

**You will need 6 TOC Pages**



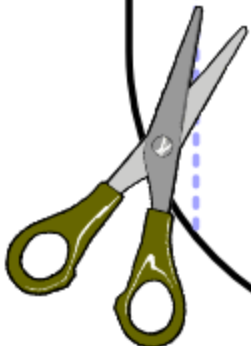
The diagram shows a rounded rectangle representing a page. A vertical black line divides it into two equal halves. A horizontal blue line is at the top of the left half. A diagonal blue line runs from the top of the vertical line to the right edge. Two light blue rectangular boxes are positioned at the bottom right: one is horizontal and sits on the bottom edge, while the other is vertical and sits on the right edge, partially overlapping the horizontal one.

Last 2 pages

Scientific Toolbox

Scientific Toolbox

**You will trim 4 ST Pages**



Starting behind the TOC, number all of the right-hand pages.

STOP when you reach the Scientific Toolbox.

# Example



23

## Moon Phases / Cycles

★ I know the moon looks different, but I don't know why or when it changes. I learned about the words full moon (all the moon shows), half moon (only half shows) and a crescent moon (only a  $\smile$  shows).

### Moon Notes

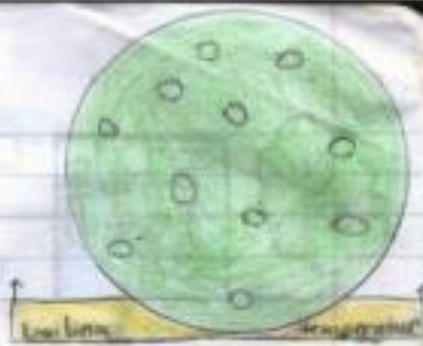
- ★ The moon rotates around the earth.
- ★ One side of the moon always faces the sun.
- ★ We see different "moons" because our position around the sun changes, which changes the light of the moon as the sun hits it.
- ★ The moon does not make (produce) its own light.
- ★ The phases or positions of the moon we see depends on where the moon, sun, and earth are.
- ★ There is a new moon (can't see it), first quarter, full moon, and third quarter (half moon).

## States of Matter

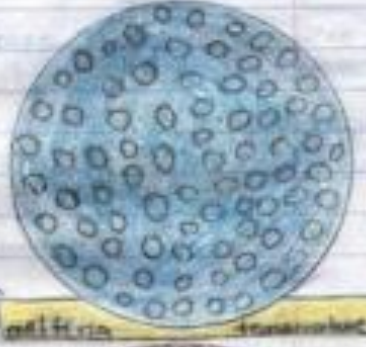
### Facts of a gas:

- ★ The particles move fast and away from each other.
- ★ The temperature is high.




### Facts of a liquid:

- ★ The particles of a substance are farther apart and slide by each other (it can flow).
- ★ The molecules move faster.
- ★ The temperature increases.
- ★ Molecules take the shape of their container.
- ★ Liquids are denser than a gas.



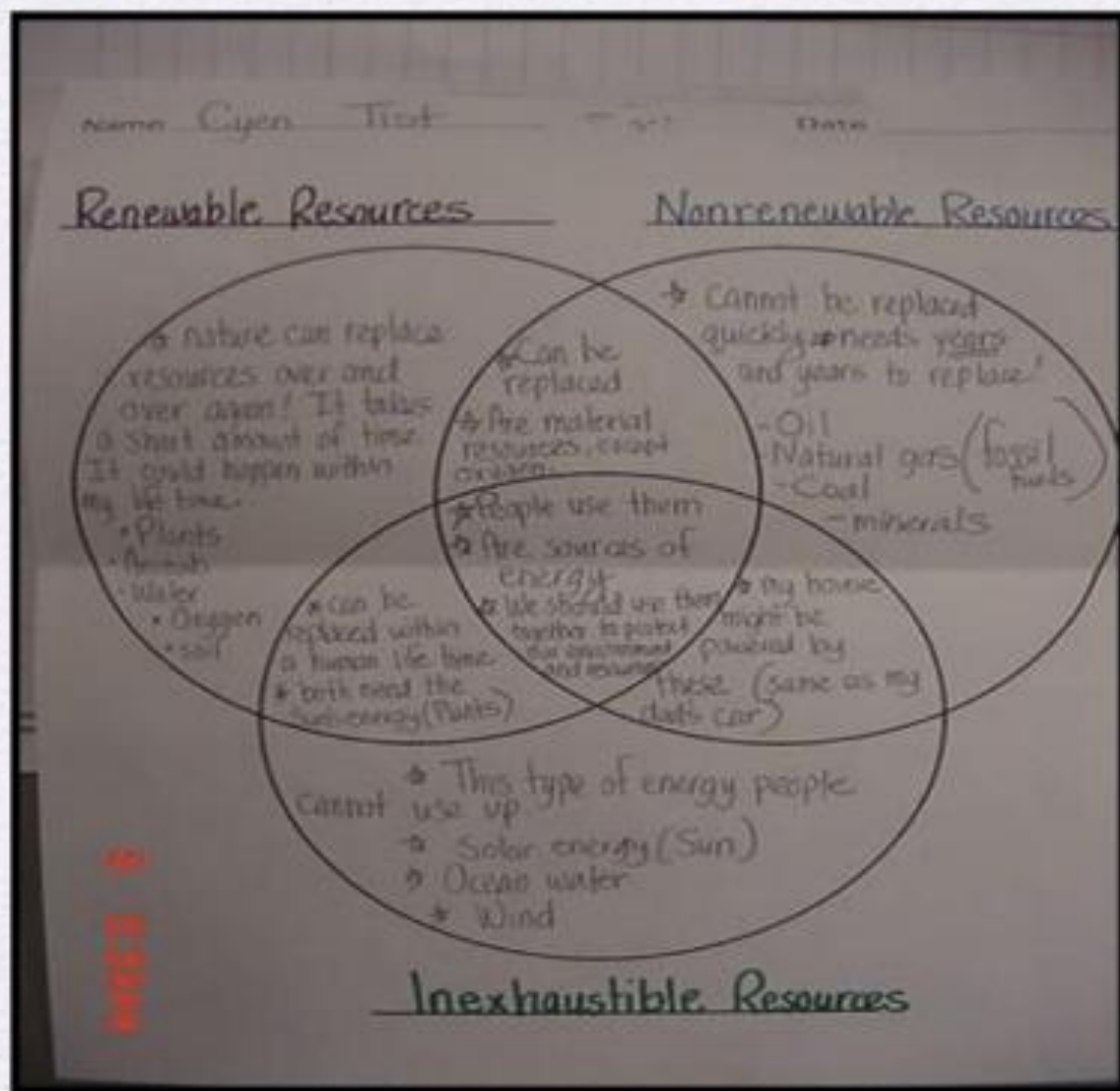
### Facts of solids:

- ★ Particles are close together.
- ★ Molecules move slow.
- ★ The temperature of the substance decreases.
- ★ The substance contracts.
- ★ A solid keeps its shape + volume.
- ★ The particles are locked together.





# Graphic Organizer



Mistakes should be  
struck out with one line!



The ~~larva~~ larvae wriggled  
atop the grain.

Information should be  
clear and understandable  
by any reader.

# Observations & Sketches

8/25/15

# IN Activity

1. How do scientists use ISNs?  
(Interactive Science Notebooks)

Team chat and write down your answer

# SCIENTIST'S NOTEBOOKS

- The following slides show real notebooks from scientists who work at Battelle – Pacific Northwest National Laboratory in Richland, Washington.
- Watch carefully as important parts to a scientist's notebook are shared.

# Computational Chemist

"Something wrong with this"

4  
1 June 96

BLYP/aug-cc-pVDZ sym. relaxed UHF

r (Å)	BLYP
1.10	-149.890988839
1.11	-149.893260068
1.115	-149.892802792
1.12	-149.893174980
1.125	-149.893433969
1.13	UHF unable to resolve instability - claims 2
1.14	-149.893611856
1.15	-149.893260068

① → 5th wrong with this

② → obviously an odd surface. Going to look @ S.C. BLYP.

③ → will have all of these check for instability and optimize.

r (Å)	BLYP1	BLYP2	charge / dipole?
1.00	-149.842029270	-149.842029270	okay
1.10	-149.890976811	-149.890976939	"
1.11	-149.892302486	-149.892302612	"
1.125	-149.892786915	-149.892787041	"
1.12	-149.893158428	-149.893158468	okay.
1.13	-149.893158468	-149.89315	"
1.13	-149.893580217	-149.893580348	BLYP3 " BLYP4
1.14	-149.893600832	-149.893600832	-149.893600832 -149.893600968
1.15	-149.893251227	-149.893251367	"

④ → x found second instability, unable to resolve it.

additional pts:

1.16	-149.892560527	-149.892560527	-149.892560527	-149.892560527
1.17	-149.891555562	-149.891555562		

⑤ → I think there's only 1 instability, but it has trouble getting there. Note identical energies for multiple instabilities.

1.18	-149.890261689	-149.890261831
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**Battelle**

... Putting Technology To Work



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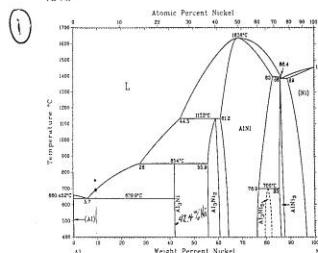
# Reference graphs and tables pasted into notebook

## Materials Scientist

9/13 B. Centrifugal Casting of Al (A356 or A390) w/ pptn. of Intermetallics.

Initially attempting an Al-Ni intermetallic.

Al-Ni



P. Nash, M.F. Singleton, and J.L. Murray, 1991

Phase	Composition, wt% Ni	Pearson symbol	Space group
(Al)	0 to 0.24	cF4	Fm $\bar{3}m$
Al <sub>3</sub> Ni	42.4	cF16	P6 <sub>3</sub> /mmc
Al <sub>2</sub> Ni	55.9 to 65.7	hP3	P6 <sub>3</sub> /mmc
AlNi	61 to 83.0	cF2	Fm $\bar{3}m$
AlNi <sub>2</sub>	79 to 82	cF4	Cmcm
AlNi <sub>3</sub>	85 to 87	cF4	Fm $\bar{3}m$
(Ni)	89.0 to 100	cF4	Fm $\bar{3}m$

$$\rho_{Al_3Ni} = 3.957 \text{ g/cc}$$

$$\rho_{Ni} = 8.9 \text{ g/cc}$$

$$\rho_{Al} = 2.7 \text{ g/cc}$$

$$\rho_{MMC} = 2.83$$

①

We want 10 vol.% Al<sub>3</sub>Ni in a 90% Aluminum Matrix.

on R.O.M. we know:

$$\rho_{MMC} = \frac{V_{Al}}{V_{Al} + V_{Al_3Ni}} \rho_{Al} + \frac{V_{Al_3Ni}}{V_{Al} + V_{Al_3Ni}} \rho_{Al_3Ni} = 90\% (2.7) + 10\% (3.96) = 2.83 \text{ g/cc } \rho_{MMC}$$

from ① on pg. 9 we know that:

$$\left( \frac{W}{10} \right)_{Ni} = \frac{(\rho_{Ni} X_{Al}) - \rho_{MMC}}{\rho_{Al} - \rho_{Ni}}$$

→ this assumes that  $\rho_{MMC}$  is correct from above.

③

$$\text{Thus } W\%Ni \text{ in the in-situ MMC is } = \frac{(8.9)(2.7) - 2.83}{2.70 - 8.9} = 0.066 \text{ or } 6.6\% Ni$$

Compare this to a simple volume to mass conversion using the density of the Al<sub>3</sub>Ni yields:

$$C = X \rho \text{ or } X = \frac{C}{\rho}$$

when  $X = \text{wt}\%$ ,  $C = \text{vol}\%$ ,  $\rho = \text{density}$

$$\text{Thus, } X_{Al_3Ni} = \frac{10}{3.957} = 2.53\% Al_3Ni$$

④

Based on a wt-fraction of Ni = 6.6% it appears that there may not be sufficient time to cool melt at this composition to retard solidification of Al before there is time to centrifugally distribute & precipitating Al<sub>3</sub>Ni. Therefore, we will recalculate the above, shooting for a 15% Al<sub>3</sub>Ni.

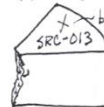
$$\rho_{MMC} = (85)(2.7) + (15)(3.957) = 2.89 \text{ g/cc}$$

$$W\%Ni = \frac{(8.9)(2.7) - 2.89}{2.70 - 8.9} = 0.0944 \text{ or } 9.4\% Ni$$

This indicates that from the melt <sup>1st</sup> solidification (of Al<sub>3</sub>Ni) occurs at approx. 700°C

3-5-96

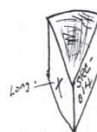
Samples Taken to L. Priest, for Metallography Prep. (+ C.E. Chamberlain)



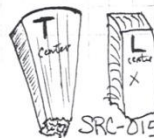
SRC-013: Al 339/5% NiAl (Un-Oxidized)

Metallog. Results: Heavy Ren between matrix and reinforcement indicative of feather-like/needle-like products.

①



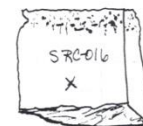
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SRC-015: Al 339/8.7% NiAl (Oxidized Powder Crossed, 24 hrs) [Remelted SRC-014]

Metallog. Results indicate distinct regions of unreinforced and reinforced matrix. Particles in outer reinforced region are both unreacted and reacted w/ matrix.

②



SRC-016: Al 339/3.2% NiAl (Oxidized NiAl powder @ 1000°C, 24 hrs) [Stock Mat'l. for SRC-017]

Metallography Results indicate distinct reinforcement mixed in matrix.

③



SRC-017: Al 339/6% NiAl (Oxidized NiAl @ 1000°C, 24 hrs) [Remelted SRC-016 PLUS AD: 10% NiAl to melt]

Same as SRC-015. Inner Region: Unreinforced Outer Region: Reinforced

See Page 36 for Copies of Metallographs



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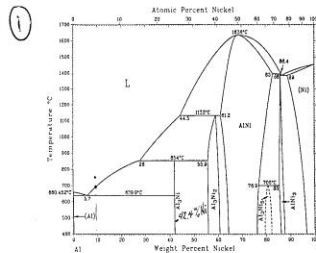
# Materials Scientist

Sample  
sketch

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$$2.83 = \frac{90}{90 + 10} (2.7) + \frac{10}{90 + 10} (3.957)$$

from ① on pg. 9 we know that:

$$\left( \frac{W}{10} \right)_{Ni} = \frac{(\rho_{Ni} X_{Al})}{\rho_{MMC}} - \rho_{Ni}$$

$$\rho_{Al} - \rho_{Ni}$$

→ this assumes that  $\rho_{MMC}$  is correct from above.

③

$$\text{Thus } \frac{W}{10} \text{ Ni in the in-situ MMC is } = \frac{(8.9)(2.7) - 8.9}{2.83 - 8.9} = 0.066 \text{ or } 6.6\% \text{ Ni}$$

Compare this to a simple volume to mass conversion using the density of the Al<sub>3</sub>Ni yields:

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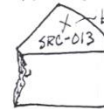
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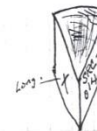
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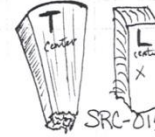
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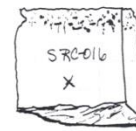
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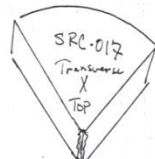
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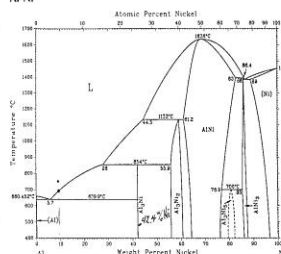
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$$C = X \rho \text{ or } X = \frac{C}{\rho}$$

when  $X = \text{wt\%}$ ,  $C = \text{vol\%}$ ,  $\rho = \text{density}$

$$\text{Thus, } X_{Al_3Ni} = \frac{10}{3.957} = 2.53 \text{ wt\% Ni}$$

④

Based on a wt-fraction of Ni = 6.6%, it appears that there may not be sufficient time to cool, melt at this composition to retard solidification of Al before there is time to centrifugally distribute & precipitating Al<sub>3</sub>Ni. Therefore, we will recalculate the above, shooting for a 15% Al<sub>3</sub>Ni.

$$\rho_{MMC} = (85)(2.7) + (15)(3.957) = 2.89 \text{ g/cc}$$

$$\frac{W}{10} \text{ Ni} = \frac{(8.9)(2.7) - 8.9}{2.89 - 8.9} = 0.0944 \text{ or } 9.4 \text{ wt\% Ni}$$

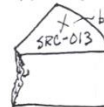
Not Valid when compared to 15% Al<sub>3</sub>Ni

$$9.4 \text{ wt\% Ni}$$

This indicates that from the melt 1<sup>st</sup> solidification (of Al<sub>3</sub>Ni) occurs at approx. 700°C

3-5-96

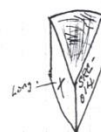
Samples Taken to L. Priest, for Metallography Prep. (+ C.E. Chamberlain)



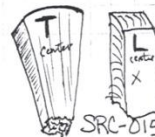
SRC-013: Al 339/5% NiAl (Un-Oxidized)

Metallog. Results: Heavy Rim between matrix and reinforcement indicative of feather-like/needle-like products.

①



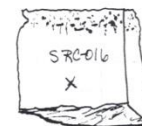
SRC-014: Al 339/8.7% NiAl (Oxidized @ 1000°C, 24 hrs) [Stock Mat'l. for SRC-015]



SRC-015: Al 339/8.7% NiAl (Oxidized Powder Cross-section, 24 hrs) [Remelted SRC-014]

Metallog. Results indicate distinct regions of unreinforced and reinforced matrix. Particles in outer reinforced region are both unreacted and reacted w/ matrix.

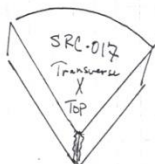
②



SRC-016: Al 339/3.2% NiAl (Oxidized NiAl powder @ 1000°C, 24 hrs) [Stock Mat'l. for SRC-017]

Metallography Results indicate distinct reinforcement mixed in matrix.

③



SRC-017: Al 339/6% NiAl (Oxidized NiAl @ 1000°C, 24 hrs) [Remelted SRC-016 PLUS AD: 10% NiAl to melt]

Same as SRC-015. Inner Region: Unreinforced Outer Region: Reinforced

See Page 36 for Copies of Metallographs

Results (crossed out)

## Description of work to date pertaining to ELK EXCLOSURES on ALE.

25

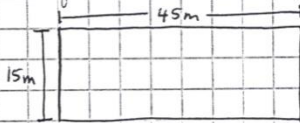
July 1997:

① Due to growing elk populations on ALE, the need to assess their impact on vegetation became apparent. Kent Johnson (FALS) and Jen Morse (AWU/SEKS) went to ALE to look for potential sites for Elk exclosures. Also they located fence posts left from old fences on the Reserve. Criteria for sites included accessibility, roads, evidence of elk presence (including trails), different vegetation types, and likelihood of continued elk presence.

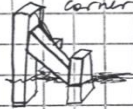
② Four sites were chosen: near Rattlesnake Springs, near the power lines at gate 120, at Upper Snively Canyon, and the last past Rattlesnake Springs off a road which is a right turn from the main road of Gate 118. (Sorry that I don't know the Road name)

August 1997.

Larry Adnell, Bob Feldman, Kent Johnson, WA Dept of Fish & Wildlife, and I (J Morse) built the four exclosures.



A railroad tie, about 3 ft in diameter and 7 ft above ground, was placed at each corner supporting posts/braces were also put in place at each corner post.



③ Metal fence posts were placed along the perimeter of the rectangle and driven into the ground. About 4 or 5 rows of barbed wire were strung around the perimeter as well. The structures seem very sturdy and durable.

October 1997

④ I went to each elk exclosure and set up 3 10m x 10m plots within the exclosure and two sets of 3 10x10 plots outside the exclosure. These control plots are in place so that vegetation inside + outside can be sampled and compared, thus evaluating the impact elk are having on ALE. The control plots were sited with attention to elk trails (tried to include part of at least one in a control plot + maintaining similar vegetation types + distribution, similar topography and slope). Also the control plots were located at about 15-25 m from the structures. ~~white~~ white fiberglass poles mark the corners of each 10x10m control plot. The vegetation in the control plots has not been sampled to date.

# ECOLOGIST

Describing  
the  
problem –  
the purpose  
of the study

Project No.

Date of Work

Entered By Jennifer Morse

Date 11/9/97

Disclosed To and Understood By



# ECOLOGIST

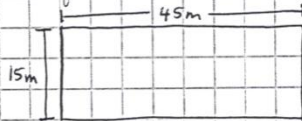
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Metal fence posts were placed along the perimeter of the rectangle and driven into the ground. About 4 or 5 rows of barbed wire were strung around the perimeter as well. The structures seem very sturdy and durable.

Identifying the site including selection criteria

October 1997

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# ECOLOGIST

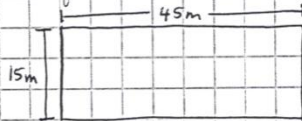
July 1997:

- ① Due to growing elk populations on ALE, the need to assess their impact on vegetation became apparent. Kent Johnson (FALS) and Jen Morse (AWU/SEKS) went to ALE to look for potential sites for Elk enclosures. Also they located fence posts left from old fences on the Reserve. Criteria for sites included accessibility, roads, evidence of elk presence (including trails), different vegetation types, and likelihood of continued elk presence.

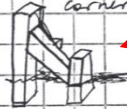
- ② Four sites were chosen: near Rattlesnake Springs, near the power lines at gate 120, at Upper Snively Canyon, and the last past Rattlesnake Springs off a road which is a right turn from the main road of Gate 118. (Sorry that I don't know the Road name)

August 1997.

Larry Adnell, Bob Feldman, Kent Johnson, WA Dept of Fish & Wildlife, and I (J Morse) built the four enclosures:



A railroad tie, about 3 ft in diameter and 7 ft above ground, was placed at each corner supporting posts/braces were also put in place at each corner post.



Metal fence posts were placed along the perimeter of the rectangle and driven into the ground. About 4 or 5 rows of barbed wire were strung around the perimeter as well. The structures seem very sturdy and durable.

Specifications regarding the Elk Enclosures

October 1997

- ④ I went to each elk enclosure and set up 3 10m x 10m plots within the enclosure and two sets of 3 10x10 plots outside the enclosure. These control plots are in place so that vegetation inside + outside can be sampled and compared, thus evaluating the impact elk are having on ALE. The control plots were sited with attention to elk trails (tried to include part of at least one in a control plot + maintaining similar vegetation types + distribution, similar topography and slope). Also the control plots were located at about 15-25 m from the structures. ~~white~~ white fiberglass poles mark the corners of each 10x10m control plot. The vegetation in the control plots has not been sampled to date.

Project No.

Date of Work

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Date 11/9/97

Disclosed To and Understood By

# THINKING ABOUT YOUR SCIENCE NOTEBOOK...

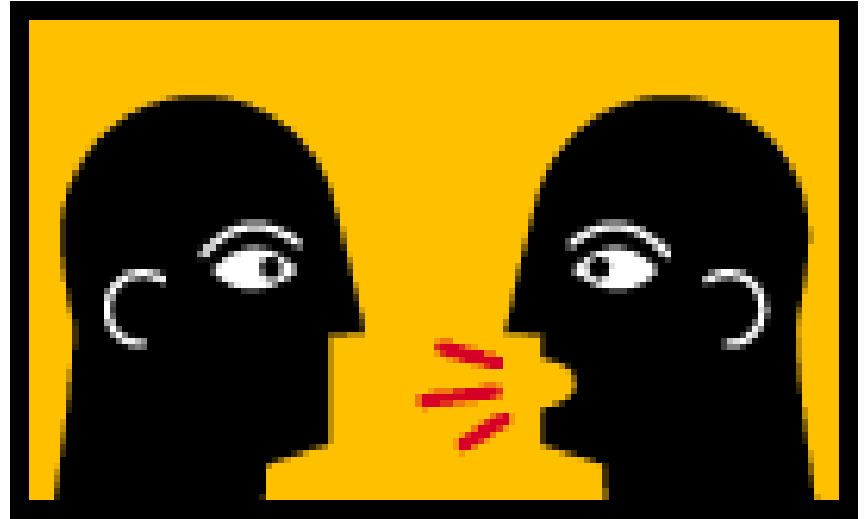
- What are some of the things you saw happening in the scientists' notebooks?

## ***OUT Activity***

*List 6 things that you observed from the scientists  
ISNs*

# THINKING ABOUT NOTEBOOKS...

Share out  
with your  
table  
group...



# REFLECT...

Which of those things do you think you could incorporate in YOUR notebook?



# SKETCH YOUR OBJECT

*Model first – describing  
what you are doing...*

Outside shape

Split object into parts



# Move from Sketch to Scientific Illustration

Label the parts

Add color and dimension

Drawing now is detailed,  
accurate, and labeled





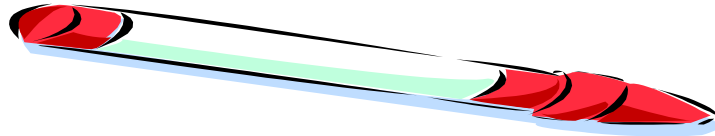
# OBSERVATIONS ORGANIZER

## *Writing Frame*

<i>Think of properties you can see such as size, shape, color, lines, texture, pattern, behavior...</i>	I observed
<i>Think of the other senses of smell, sound, touch, and perhaps taste!</i>	I noticed
<i>Connect it with something that you already know.</i>	It reminds me of
<i>Add more detail as needed.</i>	This is so because
<i>Be curious and ask questions you could investigate.</i>	I am curious about  It surprised me that  I wonder what would happen if

This organizer or writing frame is one part of a comprehensive, research-based approach to teaching students how to think, talk and write like scientists. (See *Writing in Science* by Betsy Rupp Fulwiler, © 2007, Portsmouth, NH: Heinemann.)

**Now add a new  
object...**



# THE BOX & T-CHART

Similarities

--

Object #1

Object #2

Differences

--	--

# THE BOX & T-CHART

Similarities

--	--

Object #1

Object #2

Differences

This organizer or writing frame is one part of a comprehensive, research-based approach to teaching students how to think, talk and write like scientists. (See Writing in Science by Betsy Rupp Fulwiler, © 2007, Portsmouth, NH: Heinemann.)

# COMPARE AND CONTRAST

## *Writing Frame*

<i>Start with how things are the same or similar.</i>	The _____ and the _____ are the same because they both _____.
<i>Add more details as needed.</i>	In addition, they both _____.
<i>Explain how they are different. You can compare the same property or characteristic in the same sentence. Use "and", "but", or "whereas" to set up the contrast.</i>	They are different because the _____, but the _____ does not.
<i>Add more detail as needed.</i>	Also, the _____, whereas the _____ does not.

This organizer or writing frame is one part of a comprehensive, research-based approach to teaching students how to think, talk and write like scientists. (See *Writing in Science* by Betsy Rupp Fulwiler, © 2007, Portsmouth, NH: Heinemann.)

# Update Your Table of Contents

- Turn back to your Table of Contents.
- Add “Observations and Sketches” with the corresponding date and page numbers.
- Now that you have had a basic experience in using your science notebook you are ready to record your experiences while at work in your science class!

**THEN...**

Don't forget to let your notebook  
reflect your...

**ARTISTIC  
PERSONALITY**

